FUNCTIONALIZED CHLOROENAMINES IN AMINOCYCLOPROPANE SYNTHESIS - V.¹ Synthesis and reactions of aminocyclopropylketones

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Abstract: Conjugated acylated chloroenamines **5A**,C were available by chlorination of mixtures of acylated enamines **8A**,C**/9A**,C by NCS at 0°C. Nonconjugated chloroenamines **10Aa** and **10Ba** could be obtained by NCS-chlorination of the conjugated acylenamines **9A**,B at low temperature. Reaction of **5Aa-5Ac**,**5Ae**, **5Af** with cyanide produced morpholinocyclopropylketones **11A**. In two cases aminofurans **12Ad** and **12Ca** resulted as products of this reaction. More generally amino-aryl-furans **12** were formed by thermolysis of the aryl-cyclopropylketones **11**. Amino-alkyl-furans as **12Ae** and **12Af** only could be trapped by a Diels-Alder reaction leading to **16Ae** and **16Af**. Epoxyenamines **29A**,**B** unexpectedly were produced from the interaction of cyanide with the chloroenamines **10Aa** and **10Ba**.

The reaction of carbamoylated chloroenamines 4^{1-4} with nucleophiles proved to be a useful basis for the synthesis of bicyclic lactams 2^2 or 3^1 . The formation of 2 or 3 from 4 consists of a formal combination of a ring contraction and an aza-annulation reaction. This can be effected either by a direct tandem type process¹ or by a cyclization - ring opening - aza-annulation sequence² including an aminocyclopropane derivative 1 as isolable intermediate product.



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Nu: H, CN



4: R = NHR 5: R = Aryl, Alkyl



The simple accessibility of acylated chloroenamines of type 4 from enamines 6 prompted us to look for further systems leading to the overall result of a ring contraction - hetero-annulation reaction. We synthesized, therefore, acylated chloroenamines 5 and investigated their reaction with cyanide as a typical nucleophile.

ACYLATED CHLORORNAMINES 5 AND 10

An acylation - chlorination sequence was used as access to the acylated chloroenamines 5 in analogy to the synthesis of carbamoylated chloroenamines 4^2 .

Acylation of enamines 6 by acid chlorides is a well known process⁵ leading to a mixture of isomeric acylation products 8/9 (e.g. acylation of 6A with $7a^{6} \cdot ^{7}$, $7e^{6} \cdot ^{8} \cdot ^{9}$ or $7f^{10}$, 6C with $7a^{11}$). In the case of the benzoylated derivative the conjugated isomer 9Aa is reported to predominate strongly over the nonconjugated species 8Aa in the acylation mixture⁶ · ⁷; an opposite result is described for the acetylated compound $9Ae^{6} \cdot ^{8} \cdot ^{9}$ (see also ref. ¹²). Cycloheptenyl morpholine 6B was benzoylated according to a general procedure for the acylation of enamines⁸. The product obtained in 61% yield almost exclusively consists of 9Ba (> 90%) as detected by ¹H- and ¹³C NMR spectroscopy.

Treating the crude acylation mixtures 8A,C/9A,C with N-chlorosuccinimide at 0°C gave the desired acylated chloroenamines 5A,C in 37-62% yield (calculated for 6 as starting material). The CHCl-unit in 5 unequivocally can be established by a multiplet at 4.83-4.92 ppm (5A) or 5.15 ppm (5C) in the ¹H NMR spectrum and a doublet at 54.8-55.5 ppm (5A) or 56.8 ppm (5C) in the



¹³C NMR spectrum. The observed values of the coupling constant ${}^{1}J_{CII} = 146-153$ Hz for the doublet are characteristic of an adjacent chlorine group¹³; these values exclude the existence of an isomeric compound (presence of an CHCORmoiety, Cl at the CC-double bond).

Analogous chlorination of the seven-membered system 9B with NCS at 0°C led to a nonconjugated chloroenamine 10Ba instead of the expected conjugated species 5Ba. A purer product 10Ba (53% yield) was obtained by running the chlorination reaction at -20°C. The nonconjugated chloroallyl group is indicated in the ¹³C NMR spectrum by two singlets at 151.5 ppm and 84.1 ppm and one doublet at 120.5 ppm. Similarily pure 10Aa was accessible in 41% yield from the chlorination of 9Aa with NCS at -70°C (¹³C NMR of the chloroallyl moiety: singlets at 146.3 ppm and 79.0 ppm, doublet at 109.6 ppm). Thus far 10Ba could not be converted into the conjugated isomer 5Ba. 10Ba proved to be stable for a longer period even in boiling dichloromethane solution. 10Aa, however, isomerized into 5Aa in a chloroform solution upon standing at room temperature for 24 h.



Direct NCS-chlorination transforms a nonconjugated acylenamine 8 into chloroenamine 5 and a conjugated enamine 9 into chloroenamine 10. 8, thereby, is much more reactive than 9. Benzoylated chloroenamine 5A was formed to a higher extend from a mixture 8A/9A than 8, its direct precursor, was present in this mixture. This requires at least one isomerization process which could take place at the stage of the starting materials 8/9 or the products 5/10. An equilibrium at the side of the starting materials was assumed for the explanation of the reaction of 8A/9A with azodicarboxylate generating products exclusively derived from $8A.^{12}$ An isomerization at the product side was observed to be the reason for the formation of a seven-membered carbamoylated chloroenamine $4B.^2$ The chlorination of 8Aa at different temperatures leading either to 5Aa or to 10Aa and a subsequent slow transformation of 10Aa into 5Aa demonstrate that both isomerization processes can be effective for acylated six-membered enamines. In the case of the seven-membered ring an isomerization process seems to be difficult as well for 10 as for 9.

AMINOCYCLOPROPYLKETONES 11 AND AMINOFURANS 12

Reaction of the acylated chloroenamines **5Aa-5Ac**, **5Ae** and **5Af** with cyanide gave aminobicyclohexylketones **11Aa-11Ac**, **11Ae** and **11Af** in 46-70% yield. Acetonitrile - water (1:1) or ether - water (9:1) as a solvent in the presence of methyltrioctylammonium chloride as a phase transfer catalyst proved to be the best conditions for the cyclopropane formation. The cyclopropane moiety in **11A** is established by the ¹³C NMR spectrum indicating two singlets ($\delta = 48-54$ ppm) and one doublet ($\delta = 39-40$ ppm, ¹Jcm = 170-173 Hz) in the expected regions.



Morpholine 'H NMR signals of an ABXY-type (11Ae, 11Af) or of an ABXY-type with beginning coalescence (11Aa-11Ac) (CDCl₃) give the information about the uniform endo-morpholino configuration of the obtained aminobicyclohexylketones (assignment of configuration by this method is described in ref. $^{4,14-17}$). Temperature dependency of the 'H NMR - OCH₂-signals was studied for 11Aa and 11Ae in C₆D₅NO₂ (11Aa: Hx₁ = 3.71 ppm, Hx₂ = 3.66 ppm, ²J_{BB} = ³J_{BB} = 10 Hz; Hy₁/Hy₂ = 3.92 ppm, 3.98 ppm, ²J_{BB} = 10 Hz; T_c = 333 K; 11Ae: Hx₁ = 3.26 ppm, Hx₂ = 3.31 ppm, ²J_{BB} = ³J_{BB} = 11 Hz; Hy₁/Hy₂ = 3.56 ppm, ²J_{BB} = 11 Hz; T_c = 368 K). ΔG^{+} -values of 68.1-69.3 kJ/mol (11Aa) and 75.8-76.3 kJ/mol (11Ae) were calculated for the morpholine dynamics using the well known approximation formula¹⁸ as described in ref.⁴; these values unequivocally are characteristic of a morpholino moiety in the endo-position of a bicyclohexyl system.⁴, ¹⁶

Aminofuran 12Ca (12% yield) was obtained from the analogous treatment of benzoylated chloroenamine 5Ca with cyanide besides the expected aminocyclopropylketone 11Ca (38% yield). Again the ¹³C NMR signals at 47.1 (s), 46.0 (s) and 32.4 (d, ²Jc₁ = 160 Hz) are significant of the cyclopropane moiety in 11Ca; the configuration of the bicyclic ketone 11Ca, however, is not clear (see ref.^{2,3}). In the case of the nitrobenzoyl chloroenamine **5Ad** an aminofuran **12Ad** was the sole reaction product (55% yield) which was isolated. A decomposition of aminocyclopropylketones **11** is expected to be the origin of the aminofurans **12**. Thus aminofurans **12** indeed were accessible in 44-71% yield by heating the arylsubstituted bicyclic derivatives **11Aa-11Ac** or **11Ca** to 160°C for 3 h. The formation of **12** from **11** can be understood as a homoenamine type reaction¹⁹ or a heterovinyl cyclopropane rearrangement²⁰. In each case **13** should be the intermediate of the thermal decomposition of **11**.



Ring closure and elimination of HCN are the subsequent steps on the way to the aminofurans 12. Decomposition of 11Aa to aminofuran 12Aa (77% yield) also could be achieved by heating in formic acid. The opening of the cyclopropane ring is facilitated by increasing acceptor strenght of the carbonyl group (e.g. nitrophenyl moiety in the non isolable 11Ad). Heating the alkylsubstituted cyclopropylketones 11Ae and 11Af in boiling formic acid or without a solvent gave no isolable furan derivatives. It could be demonstrated that 12Ae and 12Af are not obtained due to less thermal stability of alkylsubstituted aminofurans. Decomposition of 11Ae and 11Af in the presence of maleic anhydride led to annulated aminophthalic anhydrides 16Ae and 16Af which stem from a Diels-Alder reaction of 12Ae/12Af with 14 as a trapping reagent. 15 as the



16Ac

11Ac

primary product is not stable under the reaction conditions; a cleavage of the N,O-acetal function and elimination of water led to 16Ae and 16Af as the isolable products. Analogously a Diels-Alder product 16Ac was accessible by heating maleic anhydride (14) either with aminocyclopropylketone 11Ac (+ 48% yield) or aminofuran 12Ac (+ 61% yield). In the former case a decomposition of 11Ac to 12Ac took place prior to the cycloaddition reaction with 14. ¹H NMR and ¹³C NMR data clearly demonstrate the aminophthalic anhydride constitution 16 of the Diels-Alder products. Diels-Alder reactions with furans as dienes are well known. An aminophthalic anhydride derivative, similar to 16, was reported to be the result of a Diels-Alder reaction of 14 with aminofuran 17.²¹

Simple arylsubstituted dialkylaminofurans 17 could be synthesized by deprotonation of iminium salts 18 which were obtained from tertiary amides of B-aroylpropionic acids 19, acetic anhydride and perchloric acid.²¹ The aryl moiety in 19 is essential for the success of this synthesis, alkyl substituted aminofurans could not be prepared by this procedure.²¹



The sequence $6 \div 5 \div 11 \div 12$ represents a new way to aminocyclopropylketones 11 and to aminofurans 12. Only very few aminocyclopropylketones thus far have been reported in the literature; with one exception these compounds represent no real push-pull substituted cyclopropanes due to a deactivation of the amino moiety [e.g. 20^{22} (exception, but a very special compound), $21^{23,24}$ (N-deactivation by an ester moiety), 23^{25} and 24^{26} (N-deactivation by an acyl group), 25 (N-deactivation by incorporation of the nitrogen atom into a pyrrole- 2^{7} , an imidazole- $2^{8,29}$ or an indole- 3^{9} unit); 26^{31} (a vinylogous amide



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rather than an aminocyclopropylketone)]. 21 gave rise to a ring opening reaction in the presence of acid to produce isomerized ketones 22^{23,24}. No aminofuran formation is observed in contrast to our investigations. N-Alkoxyaminocyclopropylketones 27 are described in a recent publication; the ¹³C NMR data given, however, are not in accordance with the proposed type of compounds.³²

BPOXYENAMINES 29

The nonconjugated acylated chloroenamines 10 principally should produce the same bicyclic ketones 11 upon interaction with cyanide as their conjugated counterpart 5. Epoxyenamines 29A,B, however, were formed from sodium cyanide and chloroenamines 10Aa and 10Ba in water instead of the expected cyclopropyl ketones 11Aa and 10Ba. Better yields of 29 were obtained by running the reaction in dichloromethane and by using tetraethylammonium cyanide. The resulting epoxyenamines 29 (29A: 67% yield; 29B: 70% yield) proved to be



a mixture of two diastereomeric compounds. In both cases a pure isomer was accessible by crystallization from ether (**29A**: 20% yield; **29B**: 31% yield). The ¹³C NMR spectra are in accordance with the functional groups in **29A**, B [pure isomers: CC-double bond: **29A**: 143.7 (s), 119.8 (d); **29B**: 146.8 (s), 108.7 (d); oxirane: **29A**: 69.5 (s), 57.2 (s); **29B**: 71.2 (s), 58.4 (s)]. An additional set of ¹³C NMR signals was observed for the second diastereomeric compound in the crude reaction products.

29 should be derived from an attack of the cyanide to the carbonyl group of 10a leading to intermediate 28. The latter gives 29 by a subsequent nucleophilic substitution of chloride. The formation of 29 from 10 parallels to the reaction of bromoketone 30 with cyanide generating epoxide 31.³³ The different carbonyl activities of the two isomeric chloroenamines explain the varying reaction behaviour: 10 possesses a ketone carbonyl function which easily adds cyanide at the carbonyl group to give an epoxide. Contrarily, the vinylogous carboxamide 5 do not add cyanide at the carbonyl moiety; this allows the formation of a cyclopropane species 10 by a 1,3-ring closure reaction.

EXPERIMENTAL

¹H NMR and ¹³C NMR spectra were recorded with a Bruker AM 400 or - if not otherwise noted - a Bruker WP 200 spectrometer (TMS as internal standard). IR spectra were measured on a Perkin-Elmer 397 Infrared Spectrophotometer. Melting points are uncorrected. Microanalyses were performed with a Perkin-Elmer 240 Elemental Analyzer. N₂ as inert gas and anhydrous solvents were used for the acylation and the chlorination reactions.

(2-Morpholino-1-cyclohepten-1-yl)-phenyl-methanone (9Ba): 9Ba was prepared according to a general procedure⁸ from 9.06 g (50 mmol) of enamine 6B, 9.35 mL (55 mmol) of ethyldiisopropylamine and 5.8 mL (50 mmol) of benzoyl chloride (7a) in dichloromethane (40 mL) at 0°C. Stirring at 20°C for 2 h, working up as usual and distillation in a Kugelrohr apparatus at 178-184°C/0.03 mbar gave 8.7 g (61%) 9Ba as an yellow oil. IR (film, cm⁻¹) 1630, 1580 (C=O, C=C); ¹H NMR (CDCl₃, 400 MHz) δ 1.50-1.79 {m, 6H}, 2.37-2.49 {m, 4H}, 2.58-2.68 (m, 4H), 3.70-3.78 (m, 4H), 7.30-7.50 (m, 3H), 7.70 (d, 2H); ¹³C NMR (CDCl₃) δ 199.0 (s), 160.4 (s), 140.5 (s), 131.4 (d), 128.6 (d), 128.1 (d), 123.6 (s), 66.8 (t), 50.9 (t), 31.9 (t), 30.5 (t), 30.1 (t), 27.4 (t), 25.9 (t). Anal. Calcd for C₁₈H₂₃NO₂: C, 75.76; H, 8.12; N, 4.91. Found: C, 75.1; H, 7.93; N, 4.7.

(3-Chloro-2-morpholino-1-cyclohexen-1-yl)-phenyl-methanones 5Aa-5Ad - General Procedure: A solution of acid chloride 7 (45 mmol; 7a: 6.32 g; 7b: 6.95 g; 7c; 7.88g; 7d: 8.35 g) in chloroform (10 mL) was dropped at 0°C within 10 min to a mixture of enamine **6A** (7.53 g, 45 mmol) and triethylamine (10.0 g, 100 mmol) in chloroform (10 mL). Stirring was continued for 2 h at 20°C, then the solvent was evaporated. Extraction of the residue with ether (30 mL), filtration, washing of the residue with ether $(3 \times 5 \text{ mL})$ and evaporation of the combined ether solutions gave crude benzoylated enamines 8Aa-8Ad/ 9Aa-9Ad. N-Chlorosuccinimide (6.00 g, 45 mmol) in acetonitrile (50 mL) was added to a solution of the crude acylated enamine 8A in acetonitrile (50 mL) at 0°C within 15 min. After additional stirring at 20°C for 2 h and removal of the solvent the residue was triturated with a mixture of 70 mL of H_2 O and 30 mL of ether. In the case of **5Aa** and **5Ac** the water phase was separated and extracted with ether (2 x 20 mL). Concentration of the dried, combined ether solutions to 30 mL and storing at -18°C for 16 h gave crystalline chloroenamines 5Aa and 5Ac which were recrystallized from ether (20 mL). A similar procedure was used for 5Ab and 5Ad; here, however, a portion of the chloroenamine could be obtained as a white solid by suction prior to the separation of the ether - water mixture.

(3-Chloro-2-morpholino-1-cyclohexen-1-yl)-ethanones 5Ae and 5Af - General Procedure: A solution of acid chloride 7 (50 mmol; 7e: 3.92 g; 7f: 7.73 g) in dichloromethane (20 mL) was dropped at 0°C within 15 min to a mixture of enamine 6A (8.30 g, 50 mmol) and ethyldiisopropylamine (9.35 mL, 55 mmol) in dichloromethane (20 mL). Stirring was continued for 3 h at 20°C, then N-chlorosuccinimide (5.32 g, 40 mmol) was added at 0°C under vigorous stirring in small portions to the solution. After 30 min at 0°C the solvent was removed and the residue was extracted with ether (250 mL). Washing of the ether layer with water (3×75 mL), drying and concentration of the ether solution to 30 mL and storing at -20°C for 16 h gave chloroenamines 5Ae and 5Af which were recrystallized from ether (60 mL).

(3-Chloro-2-morpholino-1-cycloalken-3-yl)-phenyl-methanones 10Aa and 10Ba - General Procedure: N-Chlorosuccinimide (2.66 g, 20 mmol) was added in small portions to a stirred solution of 20 mmol of acylenamine (8Aa/9Aa: 5.45 g; 9Ba: 5.71 g) in dichloromethane (20 mL) at -70°C (for 8Aa/9Aa) or -20°C (for 9Ba). Continuous stirring for 1 h at this temperature, removal of the solvent at -20°C and working up as described for 5Ae/5Af gave chloroenamines 10.

Bicyclic Alkanones 11Aa-11Ac, 11Ae and 11Af - General Procedure: A mixture of acylated chloroenamine 5 (10 mmol; 5Aa: 3.06 g; 5Ab: 3.20 g; 5Ac: 3.40 g; 5Af: 3.20 g), sodium cyanide (0.49 g, 10 mmol), methyltrioctylammonium chloride (0.1 mL, 0.22 mmol), water (60 mL) and acetonitrile (60 mL) was vigorously stirred at 20° C for 16 h. The reaction of 5Ae (2.44 g, 10 mmol) analogously was run in an ether (45 mL) - water (5 mL) mixture instead of acetonitrile - water. Addition of ether (60 mL), separation of the ether phase, extraction of the water by ether (2 x 30 mL) and evaporation of the combined ether solutions gave crude 11A which was purified by recrystallization.

1 α , 5 α , 6 β -1-(4-Methylbenzoyl)-6-morpholino-bicyclo[3.1.0]hexane-6-carbonitrile (11**ab**): Crystallized from ether - pentane (1:1); yield 1.74 g (56%); mp 88°C; IR (KBr, cm⁻¹) 2230 (C-N), 1670 (C=O); ¹H NMR (CDCl₃) δ 1.80-2.13 (m), 2,13-2.48 (m), 2.38 (s) (9 H), 2.48-3.00 (m, 5H), 3.50-3.75 (m, 2H), 3.75--4.00 (m, 2H), 7.24, 7.28, 7.75, 7.79 (AA'XX'-system, 4H); ¹³C NMR (CDCl₃) δ 199.5 (s), 144.5 (s), 133.7 (s), 129.6 (d), 129.2 (d), 115.3 (s), 66.9 (t), 52.0 (s), 51.0 (t), 50.8 (t), 47.9 (s), 40.1 (d, ¹J_{CH} = 170 Hz), 30.4 (t), 27.5 (t), 25.8 (t), 21.6 (q). Anal. Calcd for C₁₉H₂₂N₂O₂: C, 73.52; H, 7.14; N, 9.03. Found: C, 73.3; H, 7.17; N, 9.0.

1a, 5a, 68-1-(4-Chlorophenyl)-6-morpholino-bicyclo[3.1.0]hexane-6-carbonitrile (11Ac): Crystallized from ether; yield: 1.52 g (46%); mp 128°C; IR (KBr, cm⁻¹) 2220 (C-N), 1660 (C=O); ¹H NMR (CDCl₃) & 1.83-2.13 (m, 3H), 2.13-2.46 (m, 3H), 2.50-3.00 (m, 5H), 3.45-3.70 (m, 2H), 3.77-4.02 (m, 2H), 7.41, 7.45, 7.77, 7.81 (AA'XX'-system, 4H); ¹³C NMR (CDCl₃) & 195.3 (s), 139.9 (s), 134.5 (s), 130.4 (d), 129.2 (d), 115.0 (s), 66.8 (t), 51.8 (s), 51.0 (t), 50.8 (t), 48.3 (s), 40.3 (d, ¹Jc₁ = 173 Hz), 30.2 (t), 27.5 (t), 25.8 (t). Anal. Calcd for $C_{18}H_{19}ClN_2O_2$: C, 65.35; H, 5.79; N, 8.47. Found: C, 65.1; H, 5.84; N, 8.5.

Reaction of Chloroenamine 5Ad with Cyanide - 4-(5,6-Dihydro-3-(4-nitro-phenyl)-4H-cyclopenta[c]furan-1-yl/-morpholine (12Ad): Crude 12Ad was obtained from 5Ad (3.15 g, 10 mmol) according to the general procedure for the reaction of cyanide with chloroenamines 5A. Washing with water (300 mL) and crystallization from ether gave deep red crystals of 12Ad. Yield: 1.72 g (55%); mp 210°C; IR (KBr, cm⁻¹) 1500 (NO₂); ¹H NMR (CDCl₃) & 2.40 (2H), 2.65 (2H), 2.70 (2H) (AA'BB'CC'-system), 3.23-3.37 (4H), 3.75-3.88 (4H) (AA'XX'-system), 7.27, 7.31, 8.05, 8.09 (AA'XX'-system); ¹³C NMR (CDCl₃) & 150.8 (s), 143.3 (s), 139.5 (s), 137.4 (s), 134.0 (s), 124.7 (d), 121.1 (d), 110.6 (s), 66.3 (t), 47.6 (t), 31.8 (t), 25.7 (t), 24.4 (t). Anal. Calcd for C₁₇H₁₈N₂O₄: C, 64.96; H, 5.77; N, 8.91. Found: C, 64.6; H, 5.79; N, 9.3.

Reaction of Chloroenamine 5Ca with Cyanide: A mixture of acylated chloroenamine 5Ca (3.90 g, 10 mmol), sodium cyanide (0.49 g, 10 mmol), methyltrioctylammonium chloride (0.1 mL, 0.22 mmol), water (60 mL) and acetonitrile (60 mL) was vigorously stirred at 20°C for 16 h. Addition of ether (60 mL), separation of the ether phase, extraction of the water by ether (2 x 30 mL), concentration of the combined ether solutions to 20 mL and storage at -20° C for 16 h gave **11Ca. 12Ca** could be obtained by further concentration of the filtrate. **11Ca** and **12Ca** were recrystallized from ether.

Thermolysis of the Aminocyclopropylketones 11 - General Procedure: Aminocyclopropylketone **11** (5.0 mmol, **11Aa**: 1.48g; **11Ab**: 1.55 g; **11Ac**: 1.66 g; **11Ca**: 1.90 g) was heated in an N₂-atmosphere to 160° C for 3 h. Distillation in a Kugelrohr apparatus and recrystallization from ether gave pure furans **12**.

4-{5,6-Dihydro-3-(4-methylphenyl)-4H-cyclopenta[c]furan-1-yl}-morpholine
(12Ab): Distilled at 164-166°C/0.001 mbar; yield: 0.84 g (59%); 'H NMR

 $\begin{array}{l} 4-\left(3-(4-Chlorophenyl)-5,6-dihydro-4H-cyclopenta[c]furan-1-yl/-morpholine \\ (12Ac): Distilled at 175-183°C; yield: 0.66 g (44%); 'H NMR (CDCl_3) & 2.32 \\ (2H), 2.60 (4H) (AA'BB'CC'-system), 3.15-3.35 (4H), 3.75-3.95 (4H) (AA'XX'-system), 7.15, 7.18, 7.23, 7.26 (AA'BB'-system, 4H); '^3C NMR (CDCl_3) & 158.9 (s), 148.8 (s), 134.9 (s), 133.0 (s), 131.7 (s), 128.9 (d), 123.5 (d), 110.3 (s), 66.5 (t), 48.6 (t), 32.1 (t), 25.1 (t), 24.1 (t). Anal. Calcd for C_{17}H_{18}ClNO_2: C, 67.21; H, 5.97; N, 4.61. Found: C, 67.1; H, 5.96; N, 4.9. \\ \end{array}$

4-(5,6,7,8,9,10,11,12-Octahydro-3-phenyl-4H-cycloundeca[c]furan-1-yl)-morpholine (12Ca): Yield: 1.15 g (65%); mp 111°C; in the ¹H NMR spectrum identicalwith the product obtained directly from the reaction of 4Ca with cyanide.

Decomposition of 11(3)a in Formic Acid - 4-(5,6-Dihydro-3-phenyl-4H-cyclopenta[c]furan-1-yl)-morpholine (12Aa): Aminocyclopropylketone 11Aa (0.30 g,1.0 mmol) was heated to 90°C for 30 min in 1 mL of formic acid. Removal of theformic acid in vacuo, addition of ether (10 mL), extraction with saturatedaqueous Na₂ CO₃-solution, removal of the solvent and recrystallization frommethanol gave pure 12Aa. Yield 0.21 g (77%); mp 97°C; in the ¹H NMR and the¹³C NMR spectra identical with the product from the thermolysis of 11Aa.

Diels-Alder Reactions with Maleic Anhydride (14): A mixture of maleic anhydride (14) (0.20 g, 2.0 mmol) and aminocyclopropylketone 11 (2.0 mmol, 11Ac: 0.66 g; 11Ae: 0.47 g; 11Af: 0.62 g) or furan 12Ac (0.60 g, 2.0 mmol) was heated for 1 h (11Ac, 12Ac: 160°C; 11Ae, 11Af: 120°C). Addition of methanol (1 mL) to the cooled reaction product, storing at 20°C for 16 h and isolation of the resulting crystals by suction gave pure Diels-Alder compounds 16.

3,5,6,7-Tetrahydro-4-morpholino-8-(4-chlorophenyl)-1H-cyclopenta[e]isobenzo-furan-1,3-dione (16Ac): Yield: 0.37 g (48%) from 11Ac, 0.47 g (61%) from 12Ac; mp 170°C; IR (KBr, cm⁻¹): 1815, 1790, 1745 (C=O); ¹H NMR (CDCl₃) δ 2.14 (2H), 2.78 (2H), 3.14 (2H) (AA'BB'CC'-system), 3.38-3.56 (4H), 3.82-4.00 (4H) (AA'XX'-system), 7.18, 7.28, 7.40, 7.50 (AA'XX'-system, 4H); ¹³C NMR (CDCl₃) δ 162.7 (s), 161.8 (s), 156.2 (s), 147.1 (s), 146.4 (s), 134.5 (s), 133.3 (s), 131.3 (s), 130.9 (d), 128.6 (d), 128.5 (s), 119.1 (s), 67.7 (t), 51.5 (t), 33.8 (t), 32.9 (t), 26.0 (t). Anal. Cald for C₂₁H₁₈ClNO4: C, 65.71; H, 4.73; N, 3.65. Found: C, 65.8; H, 4.83; N, 3.6.

3,5,6,7-Tetrahydro-8-methyl-4-morpholino-1H-cyclopenta[e]isobenzofuran-1,3--dione (16Ae): Yield: 0.22 g (38%); mp 175°C; IR (KBr, cm⁻¹): 1815, 1780, 1750 (C=O); 'H NMR (CDCl₃) & 2.19 (2H), 2.93 (2H), 3.08 (2H) (AA'BB'CC'--system), 2.50 (s, 3H), 3.26-3.41 (4H), 3.80-3.94 (4H) (AA'XX'-system); ¹³C NMR (CDCl₃) & 164.1 (s), 161.9 (s), 156.7 (s), 146.1 (s), 146.0 (s), 130.0 (s), 128.6 (s), 119.0 (s), 67.7 (t), 51.4 (t), 33.5 (t), 31.6 (t), 25.0 (t), 14.1 (q). Anal. Calcd for $C_{16}H_{17}NO_4$: C, 66.89; H, 5.96; N, 4.88. Found: C, 66.3; H, 5.98; N, 4.8.

Reaction of Chloroenamines 10Aa and 10Ba with Tetraethylammonium Cyanide – General Procedure: A solution of chloroenamine 10a (2.0 mmol, 10Aa: 0.63 g; 10Ba: 0.64 g) and tetraethylammonium cyanide (0.51 g, 3.0 mmol) in dichloromethane (15 mL) was stirred at 20°C for 16 h. Extraction with water (3 x 10

mL), evaporation of the dichloromethane and distillation of the residue in a Kugelrohr apparatus in vacuo gave 29 as a mixture of two isomers. Crystallization from ether afforded a pure isomer.

4-Morpholino-2-phenyl-1-oxa-spiro[2,5]-4-octene-2-carbonitrile (29A): Yield: Mixture of isomers: 0.40 g (67%), colorless oil distilled at 162-166°C/0.0001 mbar; pure isomer: 0.12 g (20%), mp 174°C.

Pure isomer: IR (KBr, cm⁻¹): 2205 (C-N), 1630 (C=C); ¹H NMR (CDCl₃, 400 MHz) δ 1.13-1.24 (m, 1H), 1.37-1.53 (m, 1H), 1.57-1.68 (m, 1H), 1.71-1.84 (m, 1H), 2.13-2.32 (m, 2H), 2.57-2.68 (m, 2H), 3.09-3.23 (m, 2H), 3.75-3.92 (m, 4H), 5.55 (m_c, 1H), 7.35-7.52 (m, 5H); ¹³C NMR (CDCl₃) δ 143.7 (s), 132.4 (s), 128.9 (d), 128.4 (d), 126.4 (d), 119.8 (d), 118.1 (s), 69.5 (s), 66.5 (t), 57.2 (s), 52.1 (t), 26.2 (t), 24.4 (t), 20.6 (t).

Characteristic signals of the second isomer: ¹H NMR (CDCl₃, 400 MHz) δ 1.81– -2.12 (m, 6H), 2.69–2.79 (m, 2H), 2.83–2.95 (m, 2H), 5.18 (m_c, 1H); ¹³C NMR (CDCl₃) δ 142.9 (s), 129.1 (s), 128.8 (d), 128.1 (d), 126.7 (d). 118.4 (d), 117.8 (s), 68.4 (s), 66.2 (t), 58.1 (s), 51.0 (t), 31.0 (t), 24.5 (t), 22.0 (t).

Anal. Calcd for $C_{18}H_{20}N_2O_2$: C, 72.95; H, 6.80; N, 9.45. Found for the pure isomer: C, 73.1; H, 6.80; N, 9.6. Found for the mixture of isomers: C, 72.3; H, 6.80; N, 9.4.

4-Morpholino-2-phenyl-1-oxa-spiro[2,6]-4-nonene-2-carbonitrile (29B): Yield: Mixture of isomers: 0.434 g (70%), colorless oil distilled at 180-183°C/0.04 mbar; pure isomer: 0.19 g (31%), mp 95°C;

Pure isomer: IR (KBr, cm⁻¹): 2320 (C-N), 1620 (C=C); ¹H NMR (CDCl₃, 400 MHz) δ 1.18-1.32 (m, 1H), 1.62-1.82 (m, 2H), 1.92-2.30 (m, 7H), 2.78-2.90 (m, 2H), 3.59-3.68 (m, 2H), 3.68-3.77 (m. 2H), 4.60 (mc, 1H), 7.29 (mc, 3H), 7.58 (mc, 2H); ¹³C NMR (CDCl₃) δ 146.8 (s), 130.9 (s), 129.2 (d), 127.8 (d), 126.5 (d), 118.1 (s), 108.7 (d), 71.2 (s), 66.8 (t), 58.4 (s), 49.0 (t), 35.0 (t), 27.4 (t), 26.5 (t), 25.4 (t).

Characteristic signals of the second isomer: ¹H NMR (CDCl₃, 400 MHz) δ 0.82--0.98 (m, 1H), 1.30-1.48 (m, 2H), 1.48-1.60 (m, 1H), 2.54-2.65 (m, 2H), 2.88-2.99 (m, 2H), 5.08 (m_c, 1H), 7.35 (m_c, 3H), 7.49 (m_c, 2H); ¹³C NMR (CDCl₃) δ 149.3 (s), 131.5 (s), 129.4 (d), 128.7 (d), 126.7 (d), 117.4 (s), 109.3 (d), 70.5 (s), 66.9 (t), 59.4 (s), 50.0 (t), 30.3 (t), 26.3 (t), 25.8 (t), 25.0 (t).

Anal. Calcd for $C_{19}H_{22}N_2O_2$: C, 73.52; H, 7.14; N, 9.03. Found for the pure isomer: C, 73.5; H, 7.24; N, 9.1. Found for the mixture of isomers: C, 72.8; H, 7.20; N, 8.9.

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